(SET-2) **FORMATIVE ASSESSMENT-1** CHAPTERS - 1,2,3

Name: Roll No:	Max.Ma	rks:20
I. Answer the following questions. Each carries four marks.	2 x 4	4 = 8 M
1) The rate of raise in temperature depends upon the nature of substances. Ho	w can y	/ou
prove experimentally?		
Balance the following chemical equations.		
(i) $AgCl_2 \rightarrow Ag + Cl_2$		
(ii) $Zn + HCl \rightarrow ZnCl_2 + H_2$		
(iii) $PbO + C \rightarrow Pb + CO_2$		
(iv) $CH_4 + O_2 \rightarrow CO_2 + H_2O$	•	
II. Answer the following questions briefly. Each carries two marks.3) Write the differences between real and virtual images.	2 X 2	2 = 4 M
 4) Define exothermic and endothermic reactions. If glucose powder is added to water. What happened? Identify whether it is an exothermic reaction or an er reaction? III Answer the following in one or two sonteneos. Each carries one marks 	ndother	mic
III. Answer the following in one or two sentences. Each carries one marks5) Two spherical mirrors are obtained from a same spherical substance.	. 2 X	
Which mirror has more focal length? Either M_1 or M_2 ?		M ₂
6) Define Rancidity.		
IV. Choose the correct choice and write down in the given brackets.	6 x ′	I = 6 M
7) 1 calorie =Joule	[]
A. 4.4 B. 4.1 C. 4.2 D. 4.3		
	r	,
8) Heat flows from	[]
A. Hot body to hot body B. Hot body to cold body	[]
A. Hot body to hot bodyB. Hot body to cold bodyC. Cold body to hot bodyD. Cold body to cold body	[]
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