GENERAL SCIENCE, Paper – I

2014-15

Public Exam-Guess Paper-2

(Physical Sciences) (English Version)

Time: 2½ Hours Parts A and B Maximum Marks: 50
Time: 2 hours PART-A Max. Marks: 35

Section - I $5 \times 2 = 10$

Note:

1. Answer any five questions choosing at least two from each group.

2. Each question carries two marks.

Group -A

- **1.** Write the differences between real images and virtual images.
- 2. Write the lens maker's formula and explain the terms in it.
- **3.** Explain Kirchhoff's laws on electric current and Potential difference.
- **4.** Give a few applications of Faraday's law of induction in daily life.

Group -B

- **5.** Give two important uses of washing soda and baking soda.
- **6.** What is nl^x method? How it is useful?
- **7.** Write the differences between the properties of ionic and covalent compounds.
- **8.** Write short notes on froth floatation process?

 $4 \times 1 = 4$

Note:

- 1. Answer any four questions from the following.
- 2. Each question carries one mark.
- **9.** Why do we get dew on the surface of a cold soft drink bottle kept in open air?
- **10.** Write Fermat principle.
- **11.** If we want to take a photograph of an image formed by a convex lens, where can we keep the object ?
- **12.** Why do we keep Plaster of Paris in air tight containers?
- 13. Write the valence electronic configurations of chromium and copper.

Note:

- 1. Answer any four questions choosing at least two from each group.
- 2. Each question carries four marks.

Group -A

- **15.** Suggest an experiment to prove that the rate of evaporation of a liquid depends on its surface area and vapour already present in surrounding area.
- **16.** What is total internal reflection? Explain the formation of mirage.
- 17. What is Myopia? How can we correct it? Explain with a neat diagram.
- **18.** Prove that the resultant resistance is equal to the sum of individual resistances, when three resistances are connected in series combination.

Group -B

- **19.** How many types of chemical reactions are there? Name them. Explain each with one example.
- **20.** Explain the significance of Quantum numbers in predicting the positions of an electron in an atom.
- **21.** What is a periodic property? How do the following properties change in a group and period? Explain.
 - (a) Atomic radius

- (b) Ionization energy
- (c) Electron affinity
- (d) Electro negativity.
- **22.** Distinguish between esterification and saponification reactions in organic compounds.

Section - IV

 $1 \times 5 = 5$

Note:

- 1. Answer any one question from the following.
- 2. Each question carries five marks.
- 23. Draw a neat diagram of an AC generator.
- **24.** Draw the diagrams of five d-orbitals.

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